



## Brief Communication

## Behaviour of some organophosphorus and organochlorine pesticides in potatoes during soaking in different solutions

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Accepted 8 January 2001

**Abstract**

The efficiencies of acidic solutions (radish, citric acid, ascorbic acid, acetic acid and hydrogen peroxide), neutral solutions (sodium chloride) and alkaline solution (sodium carbonate) as well as tap water in the elimination of organochlorine and organophosphorus pesticides from naturally contaminated potatoes were examined. The results indicated that acidic solutions were more effective than neutral and alkaline solutions in the elimination of the organochlorine compounds under investigation, Radish solutions eliminated pesticides completely, except *o,p'*-DDE (73.1% loss), followed by citric and ascorbic acid solutions. On the other hand, organophosphorus pesticides (pirimphos methyl, malathion and profenofos) were eliminated more by acidic, neutral and alkaline solutions than by organochlorines. The percentage of removal ranged from 98.5 to 100% for pirimphos methyl, 87.9 to 100% for malathion and 100% for profenofos. © 2001 Elsevier Science Ltd. All rights reserved.

**Keywords:** Pesticides; Potatoes; Reagents

**1. Introduction**

Pesticides comprise a variety of toxic substances, and are used in agriculture as well as indoors to kill pests (Baxter, 1988). The health risk of pesticides to humans is worsened by the fact that many of these substances have been shown to be carcinogenic and mutagenic (Murphy, 1986). Organophosphorus pesticides have a higher acute toxicity than organochlorines, but they have the advantage of being rapidly degraded in the environment. Organochlorine compounds have been banned but their residues still appear as pollutants in food as well as the environment (Rea, 1996).

Several investigators found high levels of pesticide residues after the washing and/or safety period (Kariem et al., 1991; Ramadan et al., 1992; Saleh et al., 1993). The potato crop is considered one of the most important vegetable crops in Egypt, but it is attacked by many insect species (El-Tantawy et al., 1992). This study aimed to find more efficient washing reagents for removing pesticides from potatoes naturally contaminated with pesticides, which may be used both commercially and by individual households.

**2. Material and methods***2.1. Samples*

Twenty-five kg of potatoes were collected from the local market. A representative sample of about 2 kg was examined for pesticides, and washing treatments were carried out on potato samples previously naturally contaminated with pesticides.

*2.2. Standards*

Pesticide standards of endrin, dieldrin, lindane, aldrin, heptachlor epoxide, 2,2-dichloroethylene (*o,p'*-DDE), 1,1'-dichloro-2,2'-bis (*p'*-chlorophenyl) ethylene (*p,p'*-DDE), 1-(*o'*-chlorophenyl)-1-(*p'*-chlorophenyl)-2,2'-dichloroethane (*o,p,p'*-DDD), 1,1-dichloro-2,2'-bis (*p'*-chlorophenyl)ethane (*p,p,p'*-DDD), 1-(*o'*-chlorophenyl)-1-(*p'*-chlorophenyl)-2,2,2'-trichloroethane (*o,p,p'*-DDT), 1,1,1-trichloro-2,2'-bis chlorophenyl) ethane (*p,p,p'*-DDT), pirimphos methyl, malathion and profenofos, were purchased from Chemical Service Inc. (West Chester, PA, USA).

*2.3. Chemicals and reagents*

All chemicals (citric acid, acetic acid, hydrogen peroxide, sodium chloride and sodium carbonate) used

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were obtained from E. Merck Company (Germany). Radish (garden, *Raphanus sativus*) solutions were prepared by cutting the green part into small pieces and soaking in water at 5 and 10% concentrations.

#### 2.4. Extraction and determination of pesticides

Pesticide residues were extracted from potato tubers according to the methods of the AOAC (1995). Aliquots of 1–2 µl of extract were injected into a Hewlett-Packard gas chromatography Model 5890 equipped with an Ni<sup>63</sup> electron capture detector (ECD), flame ionization detector (FID) and integrator 3392, fitted with an HP-101 capillary column (cross-linked methyl silicon Gum), 30 m×0.25 mm×0.25 µm film thickness. The oven temperature was programmed from 160 to 220°C at the rate of 5°C/min, held for 20-min injection and detector temperatures were 220 and 300°C, respectively.

#### 2.5. Treatment of contaminated potato

The contaminated intact potato samples were soaked for 10 min in (i) tap water; (ii) acidic reagents, radish, acetic acid, citric acid and ascorbic acid solutions at concentrations of 5 and 10%; (iii) neutral reagent, 5 and 10% NaCl solutions; and (iv) alkaline reagent, 5 and 10% Na<sub>2</sub>CO<sub>3</sub> solutions.

### 3. Results and discussion

The concentrations of pesticide residues in potatoes are presented in Table 1. The results indicate that the

Table 1  
Levels of pesticide residues in potato tubers<sup>a</sup>

Compound	Levels of pesticide (µg/kg±S.D.)	
	Skin	Pulp
Organophosphorus		
Pirimphos-methyl	122.9±15	4783.10±63
Malathion	41.50±10	— <sup>b</sup>
Profenofos	—	233.10±25
Organochlorines		
Endrin	20.90±7	—
Dieldrin	—	—
Lindane	9.30±3	4.50±1.5
Aldrin	1.50±0.5	1.00±0.3
Heptachlor-epoxide	13.90±5	0.70±0.1
<i>o,p'</i> -DDE	42.20±11	10.30±0.3
<i>p,p'</i> -DDE	66.70±14	15.20±5
<i>o,p'</i> -DDD	—	42.10±13
<i>p,p'</i> -DDD	—	—
<i>o,p'</i> -DDT	—	—
<i>p,p'</i> -DDT	—	—

<sup>a</sup> Values given are mean of five replicates.

<sup>b</sup> Non-detectable.

Table 2  
Behaviour of organochlorine pesticide residues during soaking in acidic solution for 10 min

Treatment	Lindane		Aldrin		Heptachlor epoxide		<i>o,p'</i> -DDE		<i>p,p'</i> -DDE		<i>o,p'</i> -DDD	
	Mean <sup>a</sup>	Reduction (%)	Mean <sup>a</sup>	Reduction (%)	Mean <sup>a</sup>	Reduction (%)	Mean <sup>a</sup>	Reduction (%)	Mean <sup>a</sup>	Reduction (%)	Mean <sup>a</sup>	Reduction (%)
Contaminated potato sample soaking	13.8±2	100	2.5±0.3	100	14.6±2.2	100	52.5±5.1	67.6	8.1±9	100	42.1±13	100
Radish solution (5%)	— <sup>b</sup>	100	—	100	—	100	16.9±5	73.1	—	100	—	100
Radish solution (10%)	—	100	—	100	—	100	14.1±3	73.1	—	100	—	100
Acetic acid solution (5%)	4.6±1.8	66.7	0.7±0.4	72	0.7±0.3	95.2	13.±3	75.2	3.2±10	96.1	1.7±0.7	96
Acetic acid solution (10%)	3±1	78.3	0.4±0.15	84	0.5±0.1	96.6	7±2	86.7	2.1±1	97.4	1.1±0.5	97.4
Citric acid solution (5%)	—	100	0.6±0.2	76	—	100	14.5	73.1	1.6±0.4	98	—	100
Citric acid solution (10%)	—	100	0.37±0.1	85.2	—	100	11.8±2	77.5	1.3±0.3	18.4	—	100
Ascorbic acid solution (5%)	—	100	0.4±0.13	84	—	100	22.3±8	57.5	6.6±2	91.9	—	100
Ascorbic acid solution (10%)	—	100	0.23±0.0	90.8	—	100	17±5	67.6	4.2±1.2	94.9	—	100
Hydrogen peroxide solution (5%)	2.6±1	81.2	0.9±0.2	64	0.8±0.2	94.5	10.3±2	80.4	—	100	—	100
Hydrogen peroxide solution (10%)	1.5±0.4	89.1	0.55±0.1	78	0.6±0.2	95.9	6.8±2	87	—	100	—	100
Tap water	12.1±3	12	2.25±0.5	10	13.2±3	9.8	51.5±9	2	79±8	3.5	4.5±1	3.9

<sup>a</sup> Mean = µg/kg±S.D. Values given are mean of five replicates.

<sup>b</sup> Non-detectable.

organophosphorus compounds pirimphos methyl, malathion and profenofos were present at higher levels than the organochlorine compounds lindane, aldrin, heptachlor epoxide and DDT derivatives in the tested sample.

It is clear that tested potato samples contained different types of pesticides. The reason for the presence of heptachlor epoxide, despite the complete cessation of use in Egypt according to the Pest Control Program (Ministry of Agriculture), might be due to the transformation of chlordane (El-Mekkawi, 1994). The contamination by DDT indicated that this insecticide is

probably still being applied despite the Ministry's recommendation. It also shows that organophosphorus compounds are widely used.

Pesticide residues in 21 species of edible vegetables and fruits collected from the Great Cairo governorate were investigated by Abou-Arab et al. (1998). They reported that potato samples contained the highest mean levels of DDT and its derivatives (0.438 µg/kg) and lindane (0.850 µg/kg).

The distribution pattern of pesticide residues in the contaminated potatoes (Table 1) indicates that lindane, aldrin, heptachlor epoxide, DDT, pirimphos methyl

Table 3  
Behaviour of organophosphorus residues during soaking in acidic solution for 10 min

Treatment	Pirimphos-methyl		Malathion		Profenofos	
	Mean <sup>a</sup>	Reduction (%)	Mean <sup>a</sup>	Reduction (%)	Mean <sup>a</sup>	Reduction (%)
Contaminated potato sample soaking	4906±35	– <sup>b</sup>	41.5±10	–	233.1±25	–
Radish solution (5%)	–	100	–	100	–	100
Radish solution (10%)	–	100	–	100	–	100
Acetic acid solution (5%)	–	100	–	100	–	100
Acetic acid solution (10%)	–	100	–	100	–	100
Citric acid solution (5%)	–	100	2.2±0.8	94.7	–	100
Citric acid solution (10%)	–	100	1.3±0.5	96.9	–	100
Ascorbic acid solution (5%)	–	100	5±2	87.9	–	100
Ascorbic acid solution (10%)	–	100	3±1	92.7	–	100
Hydrogen peroxide (5%)	–	100	4.16±2	89.9	–	100
Hydrogen peroxide (10%)	–	100	2.9±1	93	–	100
Tap water	4273±50	12.9	36.5±10	11.6	201.6±20	13.5

<sup>a</sup> Mean = µg/kg±S.D. Values given are mean of five replicates.

<sup>b</sup> Non-detectable.

Table 4  
Behaviour of pesticide residues during soaking in neutral and alkaline solutions for 10 min

Treatment	Organoclorine residues (mg/kg±S.D.)					
	Lindane		Aldrin		Heptachlor epoxide	
	Mean <sup>a</sup>	Reduction (%)	Mean <sup>a</sup>	Reduction (%)	Mean <sup>a</sup>	Reduction (%)
Contaminated potato sample	13.8±2	– <sup>b</sup>	2.5±0.3	–	14.6±2.2	–
NaCl solution (5%)	9.9±3	28.3	0.8±0.3	68	3.9±1	73.3
(10%)	8±2	42	0.6±0.2	76	2.1±0.8	85.6
Na <sub>2</sub> CO <sub>3</sub> solution (5%)	1.5±0.5	89.1	0.4±0.1	84	1±0.3	93.2
(10%)	1.1±0.3	92	0.3±0.1	88	0.7±0.1	95.2
Tap water	12.1±3	12	2.25±0.5	10	13.2±3	9.8

  

Treatment	Organophosphorus residues (mg/kg±S.D.)		
	Pirimphos-methyl		
	Mean <sup>a</sup>	Reduction (%)	Mean <sup>a</sup>
Contaminated potato sample	4906±35	–	41.5±10
NaCl solution (5%)	–	100	–
(10%)	–	100	–
Na <sub>2</sub> CO <sub>3</sub> solution (5%)	76.4±8	98.5	3.98±0.9
(10%)	60.1±5	98.8	–
Tap water	4273±50	12.9	36.5±10

<sup>a</sup> Mean = µg/kg±S.D. Values given are mean of five replicates.

<sup>b</sup> Non detectable.

and profenofos have the power to penetrate the surface of the contaminated potato into the pulp. It is known that some of the pesticide penetrates from the surface into the deep layer of plant tissues, and the penetration depends on the stability of the insecticide penetrating the skin layer.

The effect of washing procedures on pesticide residues in potatoes was studied. The data in Tables 2 and 3 showed the effect of soaking in acidic reagent solutions (5 and 10%) for 10 min. The results indicate the efficient role of washing by radish solution, citric acid, ascorbic acid, acetic acid and hydrogen peroxide in the elimination or reduction of organochlorine (lindane, aldrin, heptachlor epoxide and DDT) and organophosphorus (pirimphos methyl, malathion and profenofos) pesticides from naturally contaminated potato tubers. It was noticed that radish solutions of 5 and 10% eliminated pesticide residues completely, except for *o,p'*-DDE. Also, citric acid and ascorbic acid solutions of 5 and 10% eliminated pesticide residues of lindane, heptachlor epoxide, *o,p'*-DDD, pirimphos methyl and profenofos completely. Acetic acid and hydrogen peroxide washing came next in importance to washing with the above solutions. On the other hand, washing with tap water provided the least effective loss with the same pesticides as previously mentioned. These results agree with those obtained by Fahey et al. (1969), Powell et al. (1970), Mesallaw and Moharran (1980), Chirila and Florall (1985), Barrow et al. (1987), Gangwar and Kumar (1988), Rao et al. (1989), Sarode and Adule (1989), El-Nabarawy et al. (1992), Ismail et al. (1993), Abou-Arab et al. (1998) and Soliman (1999). These authors reported that partial removal was affected by the washing operation (water and/or acetic acid, sodium chloride).

The data in Table 4 showed the percentage rate of removal pesticides from naturally contaminated potatoes after soaking in neutral (NaCl) and alkaline (Na<sub>2</sub>CO<sub>3</sub>) solutions (5 and 10%) for 10 min. Alkaline solutions were more effective than neutral solutions in the elimination of lindane, aldrin and heptachlor epoxide. The results also indicate that both solutions have an efficient role in the elimination of total DDT and organophosphorus (pirimphos methyl, malathion and profenofos) pesticides from contaminated potatoes.

The results clearly indicate that organophosphorus pesticides are more rapidly degraded than organochlorine pesticides as expected.

The data from Tables 2–4 demonstrate that there was a gradual increase in the percentage of reduction due to the increase concentration of different washing reagents, being more efficient than tap water. These results agreed with those obtained by Lamb et al. (1968), Mesallaw and Moharran (1980), Ismail et al. (1993) and Abou-Arab et al. (1998).

From the above results, it is clear that pesticides should be applied correctly according to good agri-

cultural practice, using only the amounts recommended. Washing with water and/or detergent solutions was necessary to decrease the intake of pesticide residues. The acidic detergent solutions are more effective in the elimination of the organochlorine pesticides under investigation than alkaline and neutral solutions. Radish is the most effective acidic solution in the elimination of pesticides, followed by citric and ascorbic acid solutions.

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